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L2(1) and XA Ordering Competition in Hafnium-Based Full-Heusler Alloys Hf_2VZ ($Z = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$)

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Abstract

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Article

L2₁ and XA Ordering Competition in Hafnium-Based Full-Heusler Alloys Hf₂VZ (Z = Al, Ga, In, Tl, Si, Ge, Sn, Pb)

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Abstract: For theoretical designing of full-Heusler based spintroinc materials, people have long believed in the so-called Site Preference Rule (SPR). Very recently, according to the SPR, there are several studies on XA-type Hafnium-based Heusler alloys X₂YZ, i.e., Hf₂VAI, Hf₂CoZ (Z = Ga, In) and Hf₂CrZ (Z = Al, Ga, In). In this work, a series of Hf₂-based Heusler alloys, Hf₂VZ (Z = Al, Ga, In, Tl, Si, Ge, Sn, Pb), were selected as targets to study the site preferences of their atoms by first-principle calculations. It has been found that all of them are likely to exhibit the L2₁-type structure instead of the XA one. Furthermore, we reveal that the high values of spin-polarization of XA-type Hf₂VZ (Z = Al, Ga, In, Tl, Si, Ge, Sn, Pb) alloys have dropped dramatically when they form the L2₁-type structure. Also, we prove that the electronic, magnetic, and physics nature of these alloys are quite different, depending on the L2₁-type or XA-type structures.

Keywords: site preference; Hf-based full-Heusler compounds; first-principles study; band structures; magnetic properties; mechanical properties

1. Introduction

Heusler alloys are a noticeable class of intermetallic materials that represent as usual by the formula X₂YZ (often called full-Heusler) [1–15] or XYZ (usually named as half-Heusler) [16], where X, Y are transition-metal-element atoms and Z is a main group element. The structure of full-Heusler alloys consists of four interpenetrating fcc lattices with four equidistant sites as basis along the diagonal of the unit cell. According to the well-known Site Preference Rule (SPR) [1–15], when the valence of the X is larger than that of Y, the atomic sequence is X^A-Y^B-X^C-Z^D and the structure is the well-known L2₁ one with prototype Cu₂MnAl. Otherwise, the alloys crystallize in the so-called XA structure, where the sequence of the atoms is then X^A-X^B-Y^C-Z^D and the prototype is Hg₂CuTi. The latter alloys are also named as inverse Heusler alloys.

To the best of our knowledge, SPR has been applied extensively in the theoretical design of full-Heusler alloys and in predictions of their electronic, magnetic, and transport behavior. Some XA-type full-Heusler alloys, Mn₂CoAl [16], Ti₂MnAl [17], and Ti₂CoSi [18], were predicted to be novel spin-gapless semiconductors (SGSs) [19,20]. Furthermore, lots of XA-type full-Heusler alloys, Sc₂-, V₂-, Cr₂-, Mn₂-, Ti₂-, Zr₂-, and even Hf₂-based alloys, were revealed to be excellent half-metallic materials (HMMs) [21]. Surprisingly, one counterexample after another, including X₂CuAl [22] and Ti₂FeZ (Z = Al, Ga) alloys [6], has been reported very recently, in which these alloys show the L2₁-type structure and disobey the SPR, so that Y with more electrons enters the B sites.

In this work, a systematic theoretical work has been carried out to examine whether the conventional SPR was suitable for the Hf_2 -based highly-ordered full-Heusler alloys. To this end, the competition between the XA and L_{21} orderings of Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) full-Heusler alloys has been studied through the first-principles calculations. Our current work shows that not all the full-Heusler alloys obey the SPR, and further, the SPR can not be considered as a concise judgement principle for the structure of highly-ordered full-Heusler alloys. Remarkably, we exhibit that atomic site occupation in these Hf_2VZ alloys is decisive in determining their electronic, magnetic, and Slater-Pauling properties. For the L_{21} -type Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) full-Heusler alloys, the phase stability from the aspect of the formation energy and mechanical behaviors has also been examined in term of theory. Details of the results are shown in the following discussion.

2. Computational Details

First-principles electronic-structure calculations were performed using density function theory (DFT) implemented in the CASTEP code [23,24] according to the plane-wave pseudo-potential method. The generalized gradient approximation (GGA) [25] was adopted for the exchange-correlation functional. For the XA-type and L_{21} -type Heusler alloys Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$), a Monkhorst-Pack special k-point mesh of $9 \times 9 \times 9$ was used in the Brillouin zone integrations with a cutoff energy of 400 eV and a self-consistent field tolerance of 10^{-6} eV. The quality of the k-point separation for the band structure calculation is 0.01 \AA^{-1} .

As shown in Figure 1, the crystal structures of XA and L_{21} types Hf_2VZ Heusler alloys have been given. For the former, the two Hf atoms sitting at the two inequivalent sites, and Hf 1, Hf 2, V and Z atoms occupy the Wyckoff coordinates A (0, 0, 0), B (0.25, 0.25, 0.25), C (0.5, 0.5, 0.5) and D (0.75, 0.75, 0.75), respectively; For the latter, the two Hf atoms are essentially equivalent, and they have the same atomic environment. Hf 1, Hf 2, V and Z atoms occupy the Wyckoff coordinates A (0, 0, 0), B (0.5, 0.5, 0.5), C (0.25, 0.25, 0.25) and D (0.75, 0.75, 0.75), respectively.

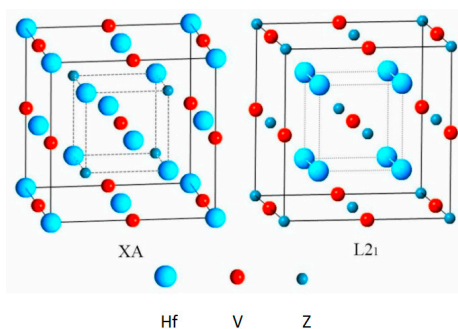


Figure 1. Crystal structures of XA and L_{21} types Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) alloys. This $1 \times 1 \times 1$ super-cell system contains 16 atoms, i.e., $8 \times \text{Hf}$, $4 \times \text{V}$ and $4 \times \text{Z}$.

3. Results and Discussion

3.1. Competition of L_{21} and XA Structure Ordering in Full-Heusler Hf_2VZ Alloys

First, to confirm the ground state of Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) full-Heusler alloys, the geometry optimization has been performed by calculating the total energies as functions of the lattice constant (cell volume) [26]. In this work, three possible magnetic states, i.e., paramagnetic (PM), ferrimagnetic (FiM), and antiferromagnetic (AFM) states are taken into account. The PM (or nonmagnetic) state means that the constituent atoms of Hf_2VZ have no spin polarization. The FiM state implies that the spin magnetic moments of Hf atoms align anti-parallel to those of the V atoms, and the total magnetic moment is not equal to zero, while the AFM state means the spin magnetic moments of Hf atoms align antiparallely to those of the V atoms, and the total magnetic

moment is equal to zero. For our Hf_2VZ ($Z = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) full-Heusler alloys, the FiM state is the most stable among three magnetic states.

As shown in Figure 2, the total energy as functions of lattice constants of full-Heusler Hf_2VZ ($Z = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) alloys with two atomic occupation orderings, XA and L2_1 , and with their most stable magnetic state, FiM, have been exhibited. Obviously, for all these alloys, the L2_1 state has lower energies than XA state. Therefore, Hf_2VZ alloys studied in current work prefer to form the L2_1 -type structure as ground state with equilibrium lattice constant of 6.69 Å, 6.66 Å, 6.90 Å, 6.89 Å, 6.56 Å, 6.61 Å, 6.82 Å and 6.90 Å, respectively. The obtained equilibrium lattice constants of these alloys with XA-type and L2_1 -type structures, respectively, and the energy differences between these two structures ($\Delta E = E_{\text{XA}}^{\text{total}} - E_{\text{L2}_1}^{\text{total}}$) for Hf_2VZ ($Z = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) alloys are also listed in Table 1. A higher value of ΔE indicates the L2_1 -type structure is more stable than XA-type. The highest positive value of 0.71 eV/cell appears in Hf_2VGa alloy, reflecting that the site preference of V for the B position is quite strong. Hence, compared to other Hf_2 -based Heusler alloys, XA-type Hf_2VGa maybe more difficult to synthesize experimentally due to its largest ΔE .

We should point out here that, as we said in the part of Introduction, Hf_2VZ alloys or even all the Hf-based full Heusler alloys should exhibit XA-type Heusler structure on basis of the SPR. Surprisingly, in current work, our results break the traditional SPR. Also, the data in this work, together with the latest scientific findings in References [6,22,27,28], are sufficient to demonstrate that not all the full-Heusler alloys X_2YZ obey the well-known SPR, especially X are low-valent transition metals. Since the SPR can not be regarded as the only way to determine the competition of XA and L2_1 structural ordering in Heusler alloys, alloys with L2_1 -type structure should also be taken into account in the previous works [1–3,7–11].

Table 1. The calculated energy difference ΔE , lattice constant a , total and atomic spin magnetic moments for Hf_2VZ alloys with L2_1 and XA structures, respectively.

| Alloy | Structure | ΔE (eV/cell) | a (Å) | M_t ($\mu_B/\text{f.u.}$) | $M_{\text{Hf } 1}$ (μ_B) | $M_{\text{Hf } 2}$ (μ_B) | M_V (μ_B) | M_Z (μ_B) | Stable Structure |
|-------------------------|---------------|----------------------|---------|-------------------------------|--------------------------------|--------------------------------|-------------------|-------------------|------------------|
| Hf_2VAl | XA | 0.64 | 6.71 | 2 | −0.65 | 0.14 | 2.62 | −0.12 | L2_1 |
| | L2_1 | | 6.69 | 1.65 | −0.21 | −0.21 | 2.19 | −0.11 | |
| Hf_2VGa | XA | 0.71 | 6.68 | 1.99 | −0.59 | 0.15 | 2.55 | −0.13 | L2_1 |
| | L2_1 | | 6.66 | 1.68 | −0.13 | −0.13 | 2.06 | −0.12 | |
| Hf_2VIn | XA | 0.39 | 6.88 | 2 | −0.75 | −0.01 | 2.88 | −0.12 | L2_1 |
| | L2_1 | | 6.90 | 1.70 | −0.24 | −0.24 | 2.31 | −0.13 | |
| Hf_2VTl | XA | 0.32 | 6.92 | 1.98 | −0.79 | −0.1 | 3.0 | −0.12 | L2_1 |
| | L2_1 | | 6.89 | 1.61 | −0.17 | −0.17 | 2.06 | −0.11 | |
| Hf_2VSi | XA | 0.63 | 6.57 | 1.02 | −0.74 | −0.25 | 2.08 | −0.06 | L2_1 |
| | L2_1 | | 6.56 | 0.24 | −0.03 | −0.03 | 0.31 | −0.01 | |
| Hf_2VGe | XA | 0.58 | 6.64 | 1.03 | −0.83 | −0.34 | 2.26 | −0.06 | L2_1 |
| | L2_1 | | 6.61 | 0.2 | −0.02 | −0.02 | 0.26 | −0.01 | |
| Hf_2VSn | XA | 0.18 | 6.85 | 1 | −1.07 | −0.51 | 2.64 | −0.05 | L2_1 |
| | L2_1 | | 6.82 | 0.49 | −0.06 | −0.06 | 0.63 | −0.03 | |
| Hf_2VPb | XA | 0.11 | 6.94 | 1 | −1.17 | −0.63 | 2.87 | −0.04 | L2_1 |
| | L2_1 | | 6.90 | 0.32 | −0.03 | −0.03 | 0.41 | −0.02 | |

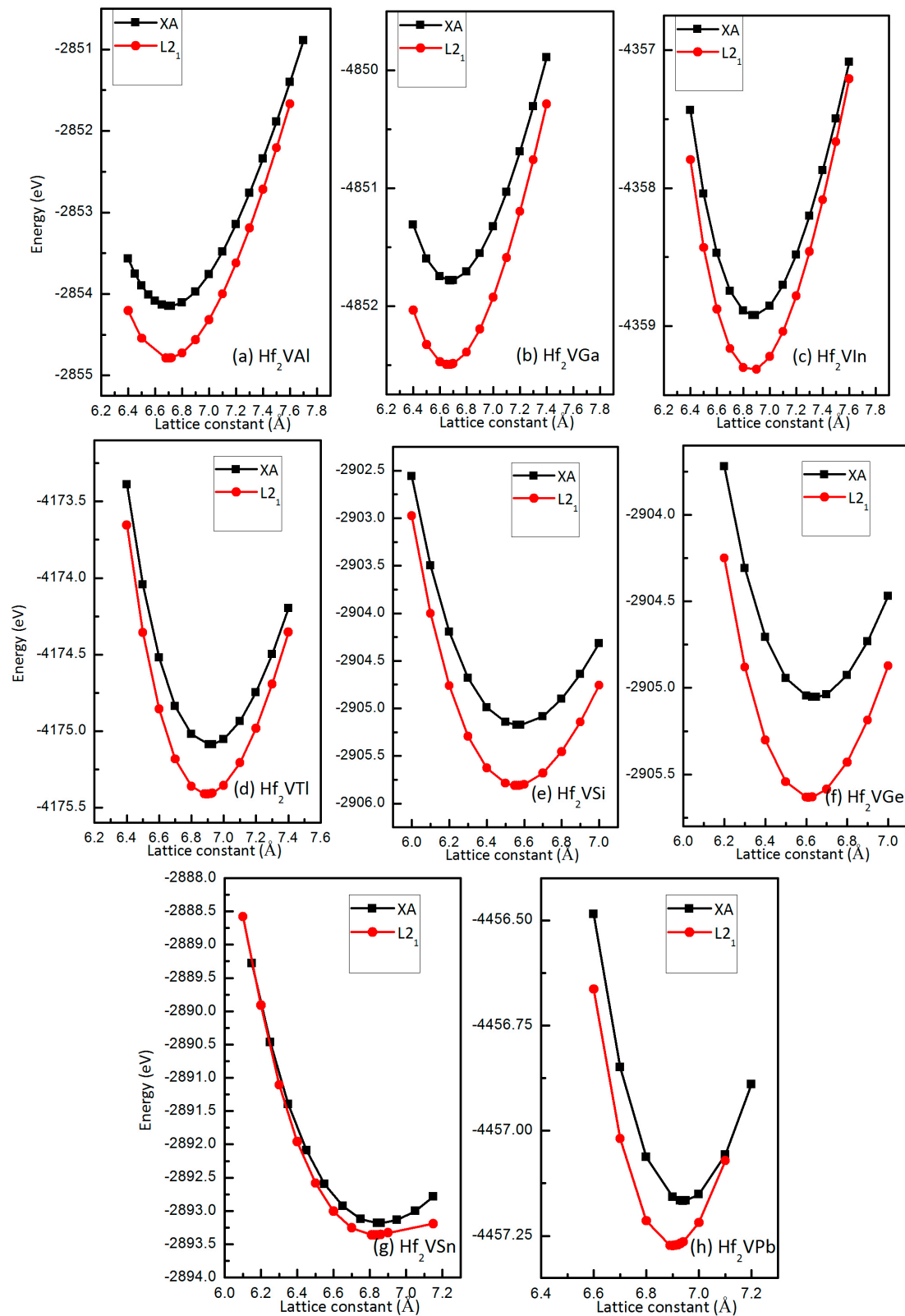


Figure 2. Calculated total energies of XA and L2₁ types Hf₂VAl (a); Hf₂VGa (b); Hf₂VIn (c); Hf₂VTl (d); Hf₂VSi (e); Hf₂VGe (f); Hf₂VSn (g); Hf₂VPb (h) alloys with respect to the lattice constant.

3.2. Structural and Mechanical Properties of Heusler-Based Hf₂VZ with L2₁-Type Ordering

Further, we aim to check the structural stability of the L2₁-type full-Heusler Hf₂VZ according to their calculated formation energies and mechanical properties. Similar methods [29,30] have been applied extensively to analyze the stability for Heusler alloys in term of theory. The formation energies of these alloys can be obtained from the following equation:

$$E_f = E_{\text{Hf}_2\text{VZ}}^{\text{total}} - (E_{\text{Hf}1}^{\text{bulk}} + E_{\text{Hf}2}^{\text{bulk}} + E_{\text{V}}^{\text{bulk}} + E_{\text{Z}}^{\text{bulk}}) \quad (1)$$

where $E_{\text{Hf}_2\text{VZ}}^{\text{total}}$ is the total energy of Hf₂VZ per formula unit, and $E_{\text{Hf}1}^{\text{bulk}}$, $E_{\text{Hf}2}^{\text{bulk}}$, $E_{\text{V}}^{\text{bulk}}$, and $E_{\text{Z}}^{\text{bulk}}$ are the total energies per atom of each element in the bulk form for the Hf 1, Hf 2, V, and Z, respectively. The results have been given in Table 2, we can see that all these alloys have negative formation energies. Furthermore, the total energy of L2₁-type Hf₂VZ is lower than XA-type Hf₂VZ (see Figure 1), and thus, the E_f of L2₁-type Hf₂VZ should be lower than XA-type Hf₂VZ. This implies that L2₁-type Hf₂VZ should be more stable than the XA-type Hf₂VZ alloys.

Next, we come to the mechanical properties of these alloys and examine their stability based on achieved elastic constants C_{ij} . For these alloys with cubic structure, only three independent elastic constants (C_{11} , C_{12} and C_{44}) are needed to be taken into consideration, and the C_{ij} can be shown as below [31]:

$$\begin{pmatrix} C_{11} & C_{12} & C_{12} & 0 & 0 & 0 \\ C_{12} & C_{11} & C_{12} & 0 & 0 & 0 \\ C_{12} & C_{12} & C_{11} & 0 & 0 & 0 \\ 0 & 0 & 0 & C_{44} & 0 & 0 \\ 0 & 0 & 0 & 0 & C_{44} & 0 \\ 0 & 0 & 0 & 0 & 0 & C_{44} \end{pmatrix} \quad (2)$$

For a small strain on a cubic system, the change of elastic energy ΔE can be shown as below [32]:

$$\Delta E = \frac{V}{2} \sum_{i=1}^6 \sum_{j=1}^6 C_{ij} e_i e_j \quad (3)$$

where V stands for the volume of the unit cell. The strain tensors are always symmetric, and they can therefore be expressed more compactly as 6-component vectors, using the so-called Voigt notation. We select three strain tensors $(0, 0, 0, \delta, \delta, \delta)$, $(\delta, \delta, 0, 0, 0, 0)$ and $(\delta, \delta, \delta, 0, 0, 0)$ to obtain the elastic constants:

$$\frac{\Delta E_1}{V} = \frac{3}{2} C_{44} \delta^2 \quad (4)$$

$$\frac{\Delta E_2}{V} = (C_{11} + C_{12}) \delta^2 \quad (5)$$

$$\frac{\Delta E_3}{V} = \frac{3}{2} (C_{11} + C_{12}) \delta^2 \quad (6)$$

where $\Delta E_1, \Delta E_2, \Delta E_3$ are the change of elastic energy for small strain tensors $(0, 0, 0, \delta, \delta, \delta)$, $(\delta, \delta, 0, 0, 0, 0)$ and $(\delta, \delta, \delta, 0, 0, 0)$, respectively.

Based on obtained C_{ij} , the mechanical properties, including bulk modulus B , shear modulus G , Voigt's shear modulus G_V , Reuss's shear modulus G_R , Young's modulus E , Pugh's ratio B/G , anisotropy factor A , of Hf₂VZ can be calculated by using the following equations [29]:

$$B = \frac{C_{11} + 2C_{12}}{3} \quad (7)$$

$$G = \frac{G_R + G_V}{2} \quad (8)$$

$$G_V = \frac{C_{11} - C_{12} + 3C_{44}}{5} \quad (9)$$

$$G_R = \frac{5(C_{11} - C_{12})C_{44}}{4C_{44} + 3(C_{11} - C_{12})} \quad (10)$$

$$E = \frac{9GB}{3B + G} \quad (11)$$

$$A = \frac{2C_{44}}{C_{11} - C_{12}} \quad (12)$$

As shown in Table 2, we can see that all these alloys with L2₁-type structure are mechanical stable due to their calculated elastic constants follow the generalized elastic stability criteria [33]:

$$C_{44} > 0 \quad (13)$$

$$\frac{(C_{11} - C_{12})}{2} > 0 \quad (14)$$

$$B > 0 \quad (15)$$

$$C_{12} < B < C_{11} \quad (16)$$

Moreover, some special mechanical properties of these alloys can also be observed from Table 2. We addressed some as follows: (1) the values of B/G of these alloys are all larger than 1.75, reflecting they are ductile according to Pugh's criteria; (2) the values of A for all these alloys are not equal to 1, meaning the fact that they are anisotropic; (3) as is known, the higher the value of E , the stiffer is the materials, and therefore, the relative stiffer order of these Hf₂VZ materials is Hf₂VAl > Hf₂VSi > Hf₂VGa > Hf₂VGe > Hf₂VIn > Hf₂VSn > Hf₂VTl > Hf₂VPb.

Table 2. Calculated elastic constants C_{ij} , bulk modulus B , shear modulus G , Young's modulus E (GPa), Pugh's ratio B/G , anisotropy factor A , and formation energy (eV) for Hf₂VZ alloys with L2₁ structure.

| Alloy | C_{11} | C_{12} | C_{44} | B | G | E | B/G | Formation Energy | Anisotropy Factor |
|---------------------|----------|----------|----------|-------|------|--------|-------|------------------|-------------------|
| Hf ₂ VAl | 164.4 | 117.6 | 73.0 | 133.2 | 46.8 | 124.6 | 2.8 | −0.54 | 3.10 |
| Hf ₂ VGa | 143 | 104.3 | 72.9 | 117.9 | 43.0 | 115.10 | 2.7 | −1.19 | 3.76 |
| Hf ₂ VIn | 179.8 | 154.3 | 49.7 | 162.8 | 28.9 | 82.0 | 5.6 | −0.05 | 3.89 |
| Hf ₂ VTl | 110.4 | 100.1 | 46.8 | 103.5 | 20.5 | 57.9 | 5.0 | −0.85 | 9.08 |
| Hf ₂ VSi | 191 | 146.9 | 63.7 | 161.6 | 41.6 | 115.12 | 3.8 | −0.66 | 2.88 |
| Hf ₂ VGe | 158.2 | 123.1 | 54.1 | 134.8 | 34.5 | 95.4 | 3.9 | −0.41 | 3.08 |
| Hf ₂ VSn | 187.5 | 155.6 | 42.2 | 166.2 | 28.5 | 81.1 | 5.8 | −0.07 | 2.64 |
| Hf ₂ VPb | 109.1 | 97.9 | 18.2 | 101.6 | 11.4 | 32.9 | 8.9 | −0.91 | 3.25 |

3.3. Calculated Electronic Behaviors of L2₁ and XA Types Hf₂VZ

Electronic-structure calculation theory has clearly indicated [20,34,35] that the spintronic properties of materials among Heusler family has an unusual sensibility to the atomic occupation in crystal cell. Hence, in this section, a discussion about the spintronic property differences, including the magnetic, electronic, half-metallic, and the spin polarization ratio (p), between the XA and L2₁ Hf₂VZ ($Z = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) full-Heusler alloys should be performed. Figures 3–5 show the calculated band structures for XA and L2₁ types Hf₂VZ ($Z = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) at their equilibrium lattice constants.

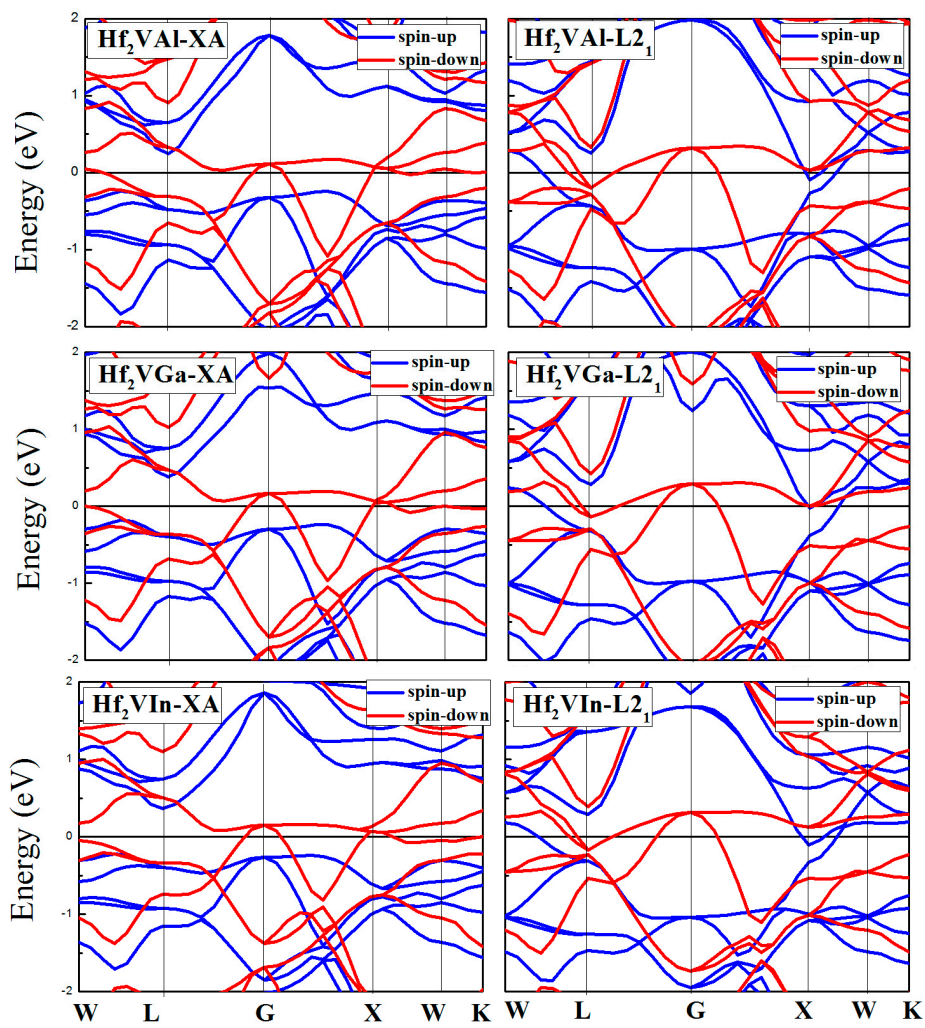


Figure 3. Calculated band structures of XA and L2₁ types Hf₂VZ (Z = Al, Ga, In) alloys.

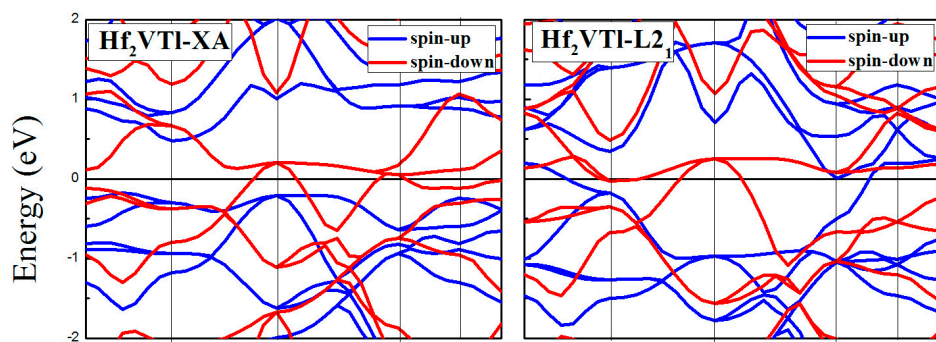


Figure 4. Cont.

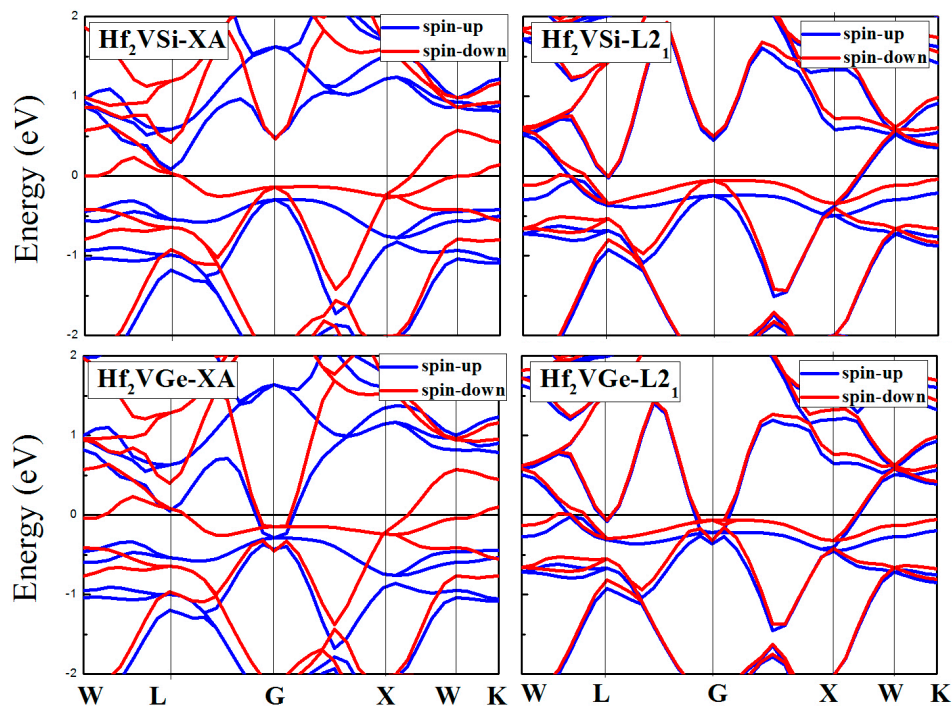


Figure 4. Calculated band structures of XA and L2₁ types Hf₂VZ (Z = Ti, Si, Ge) alloys.

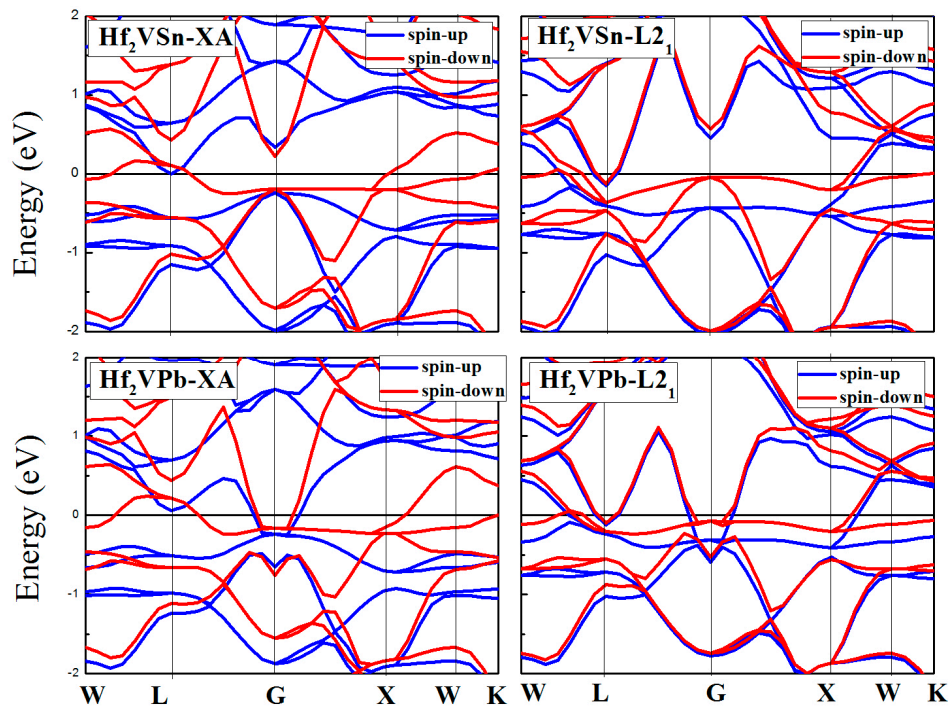


Figure 5. Calculated band structures of XA and L2₁ types Hf₂VZ (Z = Sn and Pb) alloys.

For XA atomic ordering, Hf₂VZ (Z = Al, Ga, In, Tl, Si, Sn) alloys are excellent half-metallic materials since there is a semiconducting-type band gap in the spin-up direction and the Fermi level locates between the opened gap. However, in the spin-down direction, the semiconducting-type band gap disappeared: although an opened gap can be observed near the Fermi level, the Fermi level has overlapped with the spin-down bands in varying degrees. In the spin-up channel, the significant factors, including the

calculated valence band maximum (VBM), conduction band minimum (CBM), semiconducting band gap, and spin-flip/half-metallic gap, have been given in Table 3, the semiconducting band gap [36] is the sum of the absolute values of CBM and VBM, and the spin-flip/half-metallic gap [36] is defined as the minimum value of the two absolute values of CBM and VBM. The semiconducting band gap values of these alloys are 0.46 eV for Hf₂VAl, 0.59 eV for Hf₂VGa, 0.59 eV for Hf₂VIn, 0.64 eV for Hf₂VTl, 0.35 eV for Hf₂VSi, 0.30 eV for Hf₂VSn, respectively. The spin-flip/half-metallic gap values of these alloys are 0.23 eV for Hf₂VAl, 0.23 eV for Hf₂VGa, 0.22 eV for Hf₂VIn, 0.17 eV for Hf₂VTl, 0.07 eV for Hf₂VSi, 0.04 eV for Hf₂VSn, respectively. For XA atom ordering Hf₂VGe and Hf₂VPb alloys, semiconducting band gaps in the both directions disappeared and both alloys exhibit common metallic properties.

Table 3. Calculated valence band maximum (VBM), conduction band minimum (CBM), semiconducting band gaps, spin-flip gap/half-metallic gaps and spin polarization ratio (*p*) of Hf₂VZ alloys with L2₁ and XA structures, respectively.

| Alloy | Structure | CBM | VBM | Band Gap | Half-Metallic Gap | <i>p</i> (%) |
|---------------------|-----------------|------|-------|----------|-------------------|--------------|
| Hf ₂ VAl | XA | 0.23 | −0.23 | 0.46 | 0.23 | 100 |
| | L2 ₁ | - | - | - | - | 48 |
| Hf ₂ VGa | XA | 0.36 | −0.23 | 0.59 | 0.23 | 100 |
| | L2 ₁ | - | - | - | - | 52 |
| Hf ₂ VIn | XA | 0.37 | −0.22 | 0.59 | 0.22 | 100 |
| | L2 ₁ | - | - | - | - | 36 |
| Hf ₂ VTl | XA | 0.47 | −0.17 | 0.64 | 0.17 | 100 |
| | L2 ₁ | - | - | - | - | 40 |
| Hf ₂ VSi | XA | 0.07 | −0.28 | 0.35 | 0.07 | 100 |
| | L2 ₁ | - | - | - | - | 39 |
| Hf ₂ VGe | XA | - | - | - | - | 88 |
| | L2 ₁ | - | - | - | - | 33 |
| Hf ₂ VSn | XA | 0.04 | −0.26 | 0.30 | 0.04 | 100 |
| | L2 ₁ | - | - | - | - | 56 |
| Hf ₂ VPb | XA | - | - | - | - | 94 |
| | L2 ₁ | - | - | - | - | 36 |

For Hf₂VZ (Z = Al, Ga, In, Tl, Si, Sn) alloys with the XA structure, the V and Hf atoms occupy sites with same symmetry in XA-type Heusler structure, and the hybridization of their d orbitals creates 5 bonding bands ($3t_{2g}$ and $2e_g$) and 5 non-bonding bands ($2e_u$ and $3t_{1u}$). Then, the 5 V-Hf bonding d hybrids hybridize in turn with d orbitals of Hf, again forming bonding and anti-bonding bands, while the 5 non-bonding bands ($2e_u$ and $3t_{1u}$) still exist with no hybridizing. Finally, the distribution of the 15 d orbitals in the minority-spin direction can be determined, i.e., $3t_{2g}$, $2e_g$, $2e_u$, $3t_{1u}$, $3t_{2g}$, and $2e_g$, from the high-energy level to the low-energy level. Also, we cannot ignore that Z creates 1s and 3p bands which are totally occupied in Hf₂VZ and are also below the above-mentioned 15 d orbitals. Therefore, their semiconducting-type band gaps are created by the separated Γ_{15} and Γ_{25} states, coming from the bonding t_{2g} and antibonding t_{1u} states.

The calculated total and partial density of states (TDOS and PDOS) for the six Hf₂VZ (Z = Al, Ga, In, Tl, Si, Sn) alloys with XA and L2₁ atomic orderings and for equilibrium lattice constants have also been calculated in this work to deepen the understanding of their electronic properties. As an example, the results of Hf₂VAl and Hf₂VSi alloys are given in Figure 6. Clearly, from the figure, a semiconducting band gap can be found in spin-up direction for XA-type Hf₂VZ (Z = Al, Ga, In, Tl, Si, Sn) alloys, which is in a good agreement with above-mentioned band structures.

The DOS can be widely used to analyze the bonding/anti-bonding states and the gap formation and similar analytical approach can be observed in References [19,29]. In the spin-up channel, the main peaks of Hf 2 and V atoms occurred at around −0.5 eV. In the spin-down channel, in the similar

energy region (around -0.5 eV), for V and Hf 2 atoms, such hybridized peaks appeared at the same time. Therefore, the hybridization between the V and Hf 2 atoms that formed strong bonding states at around -0.5 eV. Above the Fermi level, in the spin-up channel, the anti-bonding peak can be found at around 1 eV mainly arise from the Hf 1-d electrons, and in the spin-down channel, no opposite energy states are observed. Moreover, in the spin-up channel, the corresponding bonding-antibonding states led to the formation of an opened band gap, and the Fermi level, exactly, locates between the gap.

For L_{21} type atomic ordering, the band structures and the DOS have a big difference with XA type atomic ordering. Namely, all the Hf_2VZ alloys investigated in this work show conventional metallic behaviors without semiconducting-type band gaps at Fermi level in both spin channels. As shown in Figures 3–6, both spin-up and spin-down bands are crossed by the Fermi level.

Moreover, from the obtained total DOS, we calculated the spin polarization (p) at Fermi energy of XA and L_{21} types Hf_2VZ . The spin polarization p (%) that can be defined as the ratio of the difference to sum of the DOS values of spin up and spin down version at the Fermi-level [13,16], represented by mathematical formulation as a percent;

$$p = \frac{n \uparrow (E_f) - n \downarrow (E_f)}{n \uparrow (E_f) + n \downarrow (E_f)} \times 100\% \quad (17)$$

where the $n \uparrow (E_f)$ and $n \downarrow (E_f)$ stand for the spin-dependent DOS around the Fermi level. The results have been given in Table 3, we can see that the p of XA-type Hf_2VZ are quite high ($>88\%$), even some alloys (such as Hf_2VAl) have completely spin polarization (100%). High-spin-polarization materials are very useful in spintronic application. However, the L_{21} -type Hf_2VZ exhibit pretty low spin polarization ($<56\%$). This also implies that the novel spintronic properties (such as half-metallic properties) in XA type structure are disappeared as these alloys form L_{21} type structure.

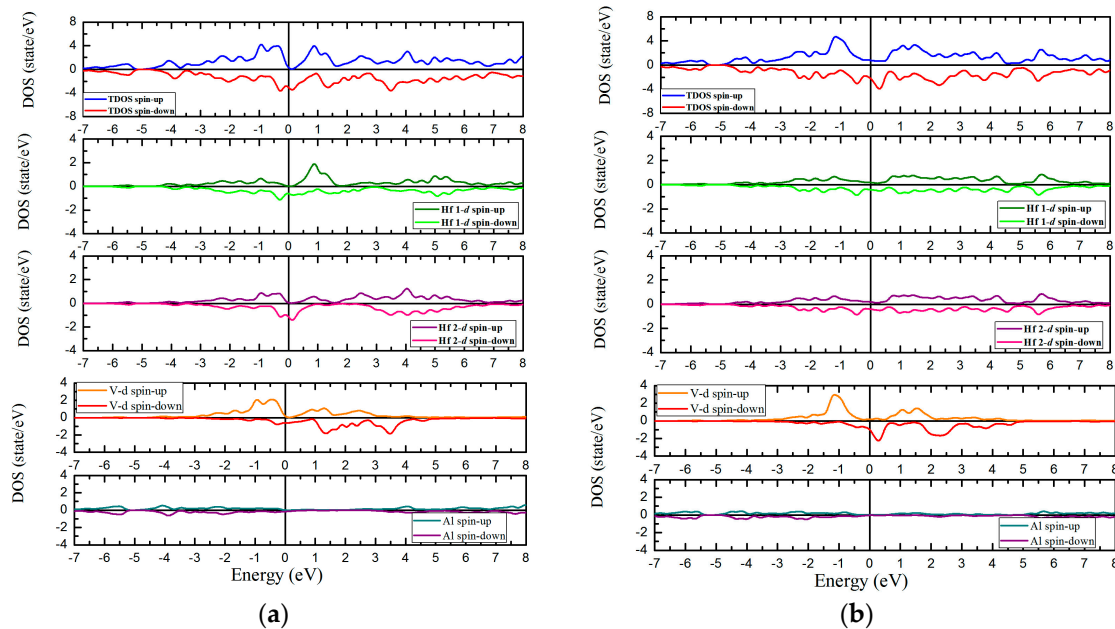


Figure 6. Cont.

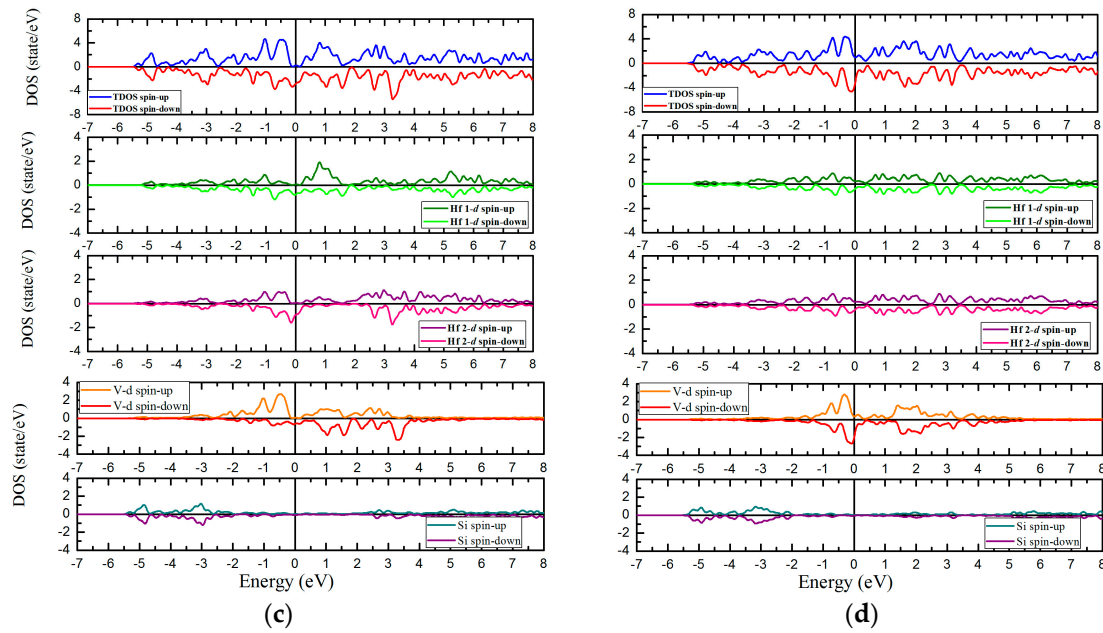


Figure 6. Calculated band structures of XA type Hf_2VAl (a); L_{21} type Hf_2VAl (b); XA type Hf_2VSi (c); L_{21} type Hf_2VSi (d); respectively.

3.4. Magnetic and Slater-Pauling Properties of L_{21} and XA Types Hf_2VZ

Finally, in this section, we come to discuss the magnetism and Slater-Pauling rule of L_{21} and XA types Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Ge, Sn, Pb}$) alloys. The total and atomic magnetic moments of these alloy with XA and L_{21} atomic ordering are shown in Table 1. Obviously, all these alloys, with either XA or L_{21} type structures, exhibit FiM properties with the atomic magnetic moments of Hf antiparallel aligned with that of V atoms. Based on Table 1, V carries the largest moment and therefore, V atom makes the most contribution to the total magnetic moment.

For XA-type Hf_2VZ full-Heusler materials, their total magnetic moments are almost integer values ($2 \mu_B/\text{f.u.}$ or $1 \mu_B/\text{f.u.}$), reflecting their high spin polarization properties. Furthermore, from Table 1, we can see that their total magnetic moments follow the well-known Slater-Pauling rule [37]:

$$M_t = Z_t - 18 \quad (18)$$

where the M_t stands for the total magnetic moment of these materials and the Z_t means the number of total of valence electrons in Hf_2VZ alloys. For L_{21} type Hf_2VZ full-Heusler materials, their total magnetic moments are all largely deviated from integer values, indicating the half-metallic properties have been totally destroyed when these alloys are in L_{21} type structure. For Hf_2VSi and Hf_2VGe , both alloys have a very weak moment of $0.24 \mu_B/\text{f.u.}$ and $0.20 \mu_B/\text{f.u.}$, respectively. Therefore, the L_{21} type Hf_2VZ full-Heusler alloys do not obey the above-mentioned Slater-Pauling rule.

The magnetism for XA and L_{21} types full-Heusler alloys Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Sn}$) alloys at their strained lattice constants has also been examined. As an example, the results of Hf_2VAl and Hf_2VGa have been given in Figure 7. According to the Slater-Pauling and generalized electron-filling rules [11,13], the integer values of total magnetic moments which follow the Slater-Pauling rule indicate the half-metallic properties of materials and thus stand for their high spin polarization. From Figure 7, we can see that the high spin polarization properties of XA type Hf_2VZ ($\text{Z} = \text{Al, Ga, In, Tl, Si, Sn}$) alloys are very robust, however, the region with high spin polarization (integer values of total magnetic moments) of L_{21} -type Hf_2VZ alloys is extraordinarily narrow. For the atomic magnetic moments of XA and L_{21} types Hf_2VZ , the values of Hf decreases, whereas the V atom, it increases.

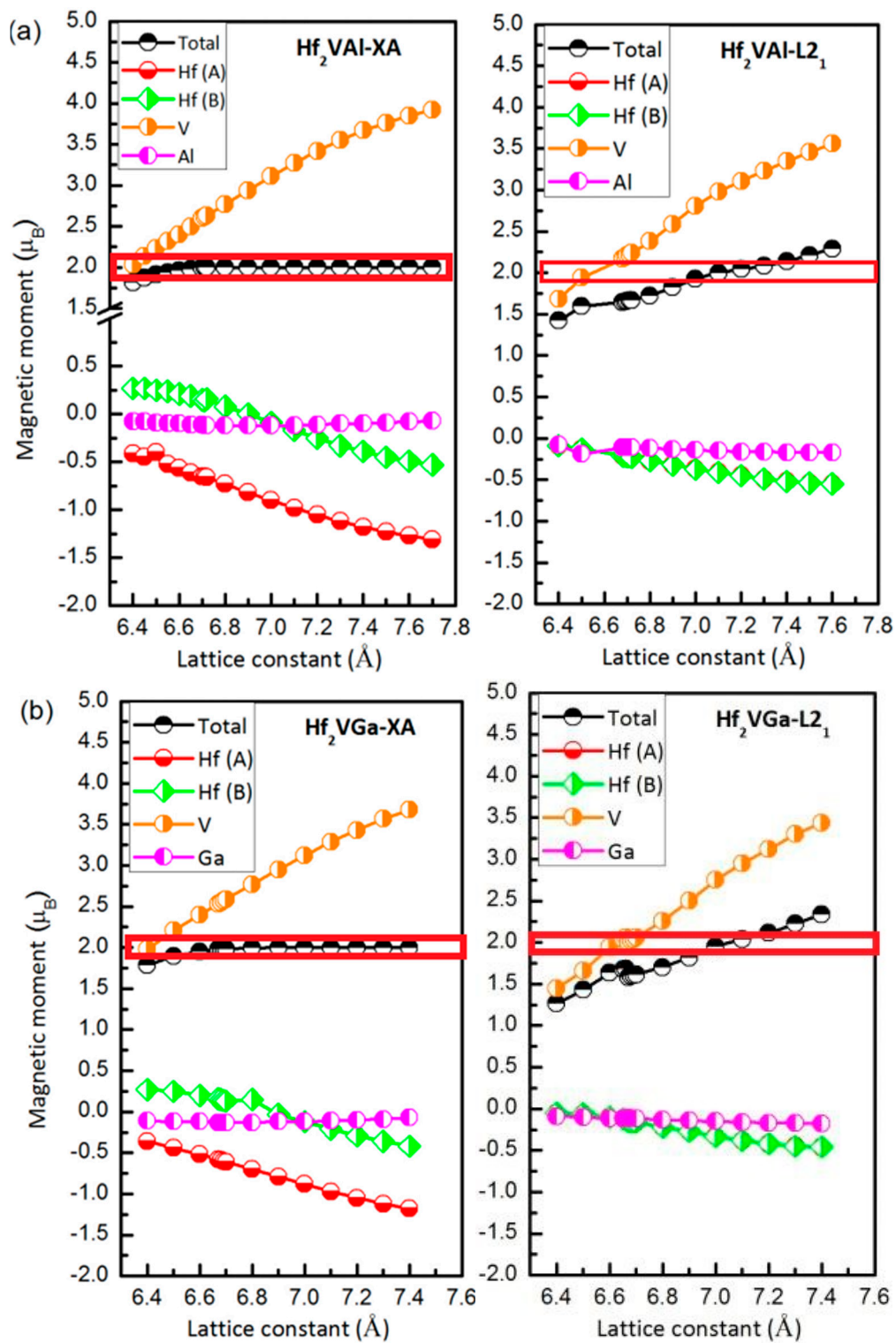


Figure 7. (a,b) Calculated total and atomic spin magnetic moments of Hf_2VZ ($Z = \text{Al, Ga}$) alloys as functions of the lattice constant. The red boxes show the high-spin polarization regions.

4. Conclusions

To sum up, the atomic occupation of a series of Hf_2 -based full-Heusler alloys was investigated by theoretical first-principle calculations. We observed that all Hf_2V -based alloys studied in current work are likely to form the L2_1 structure instead of XA structure. Our results also indicate that the spintronic

properties (half-metallic or high spin-polarization properties), and Slater-Pauling behaviors that exist in XA for Hf₂VZ are fully destroyed in L2₁. With our study, it is clear that not all full-Heusler alloys X₂YZ obey the SPR, especially when X are low valence metal elements.

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